

# Cu(II), Mn(II) and Zn(II) complexes of hydrazones with quaternary ammonium moiety: Synthesis, characterization and DFT calculation

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The **HLCl** ligand with metal salts  $\text{Cu}(\text{BF}_4)_2 \cdot 6\text{H}_2\text{O}$  /  $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$  /  $\text{Zn}(\text{BF}_4)_2 \cdot 6\text{H}_2\text{O}$  and  $\text{NaN}_3$ , in methanol form mononuclear penta coordinated complexes  $[\text{CuL}(\text{N}_3)(\text{CH}_3\text{OH})]\text{BF}_4$  (**1**) and  $[\text{ZnL}(\text{N}_3)_2]$  (**2**) and binuclear  $[\text{Mn}_2\text{L}_2(\mu_{-1,1}\text{-N}_3)_2(\text{N}_3)_2] \cdot 2\text{CH}_3\text{OH}$  (**3**) complex.

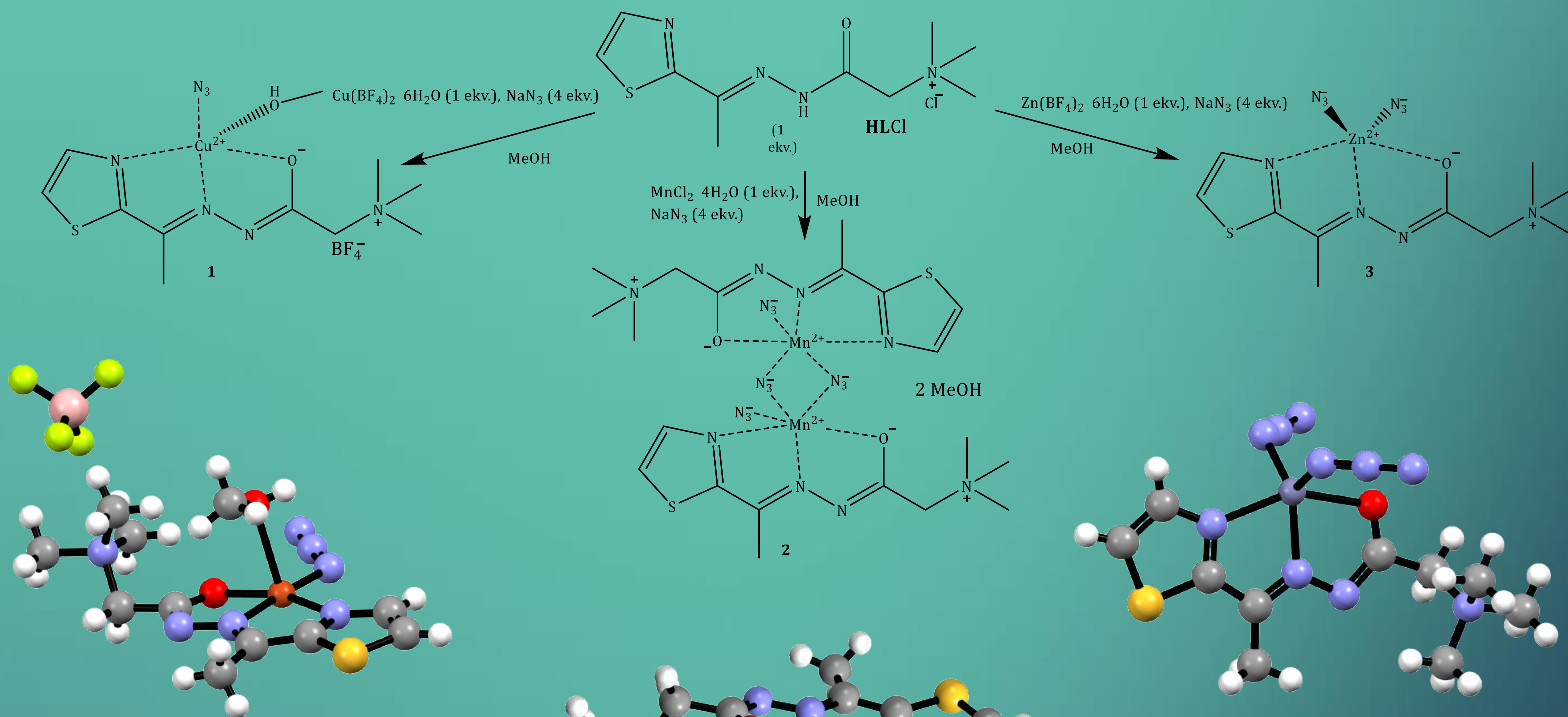


Fig. 1  $[\text{CuL}(\text{N}_3)(\text{CH}_3\text{OH})]\text{BF}_4$  complex

Fig. 3  $[\text{ZnL}(\text{N}_3)_2]$  complex

Fig. 2  $[\text{Mn}_2\text{L}_2(\mu_{-1,1}\text{-N}_3)_2(\text{N}_3)_2] \cdot 2\text{CH}_3\text{OH}$

Complexes **1**, **2** and **3** were characterized by elemental analysis, IR spectroscopy, single-crystal X-ray diffraction, and DFT calculations. In all three complexes ligand (**L**<sup>1</sup>) is coordinated in deprotonated formally neutral form *via* NNO donor set atoms. Complexes **1** and **3** crystallize in the monoclinic crystal system with space group No. 14 ( $P2_1/n$  and  $P2_1/c$  cell settings, respectively) and complex **2** in the triclinic crystal system with space group  $P\bar{1}$  (No. 2).

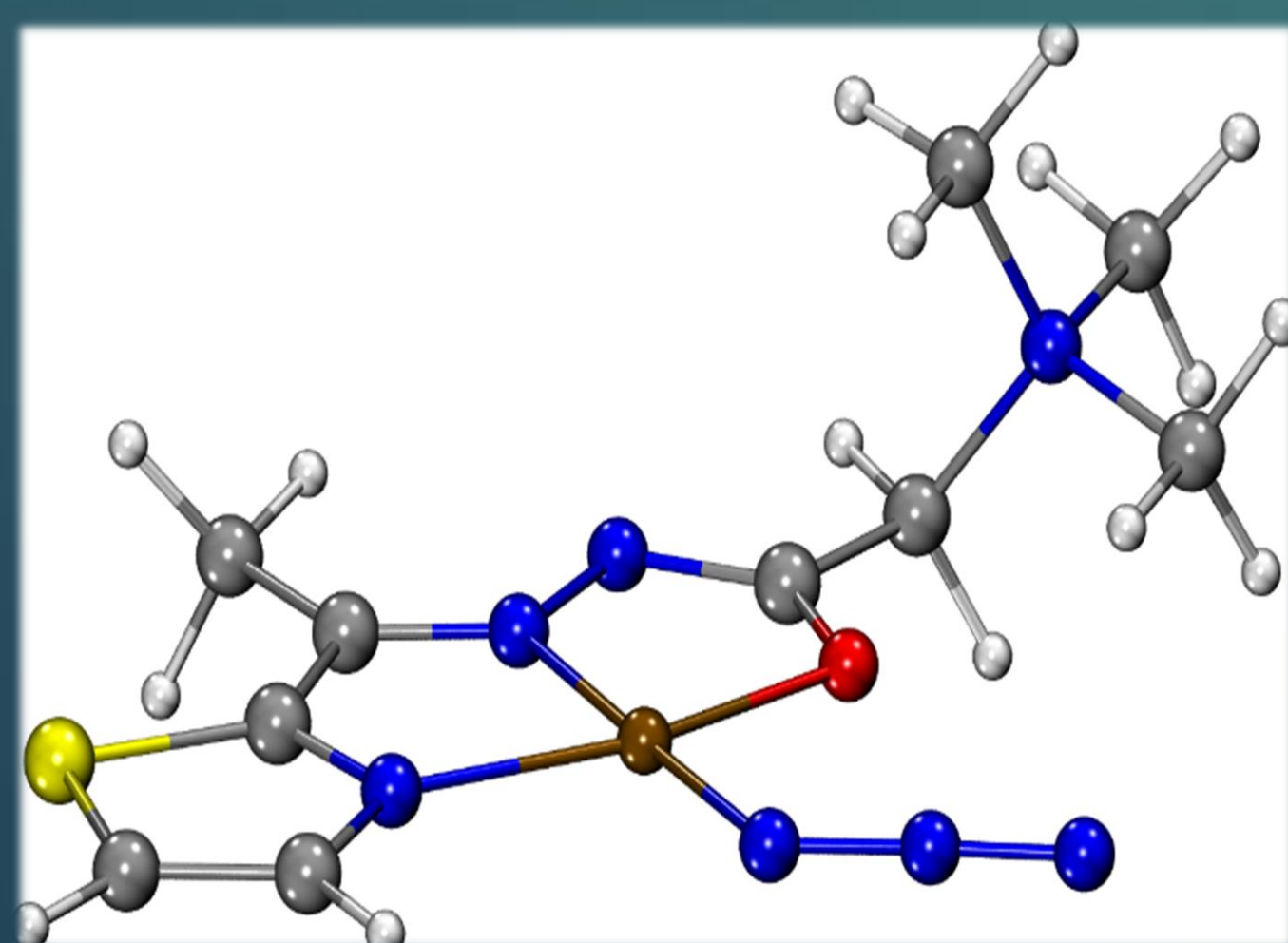


Fig. 4 Structure of  $[\text{CuL}^1(\text{N}_3)]^+$  complex ion optimized at ZORA-BP86-D3/TZP-COSMO(DMSO) level of theory

According to the DFT studies, Cu(II) complex is the most stable in square-planar geometry, while in the same DMSO solution, Mn(II) complex is the mixture of  $[\text{MnL}(\text{N}_3)_2]$  and  $[\text{Mn}_2\text{L}_2(\mu_{-1,1}\text{-N}_3)_2(\text{N}_3)_2]$  complexes.

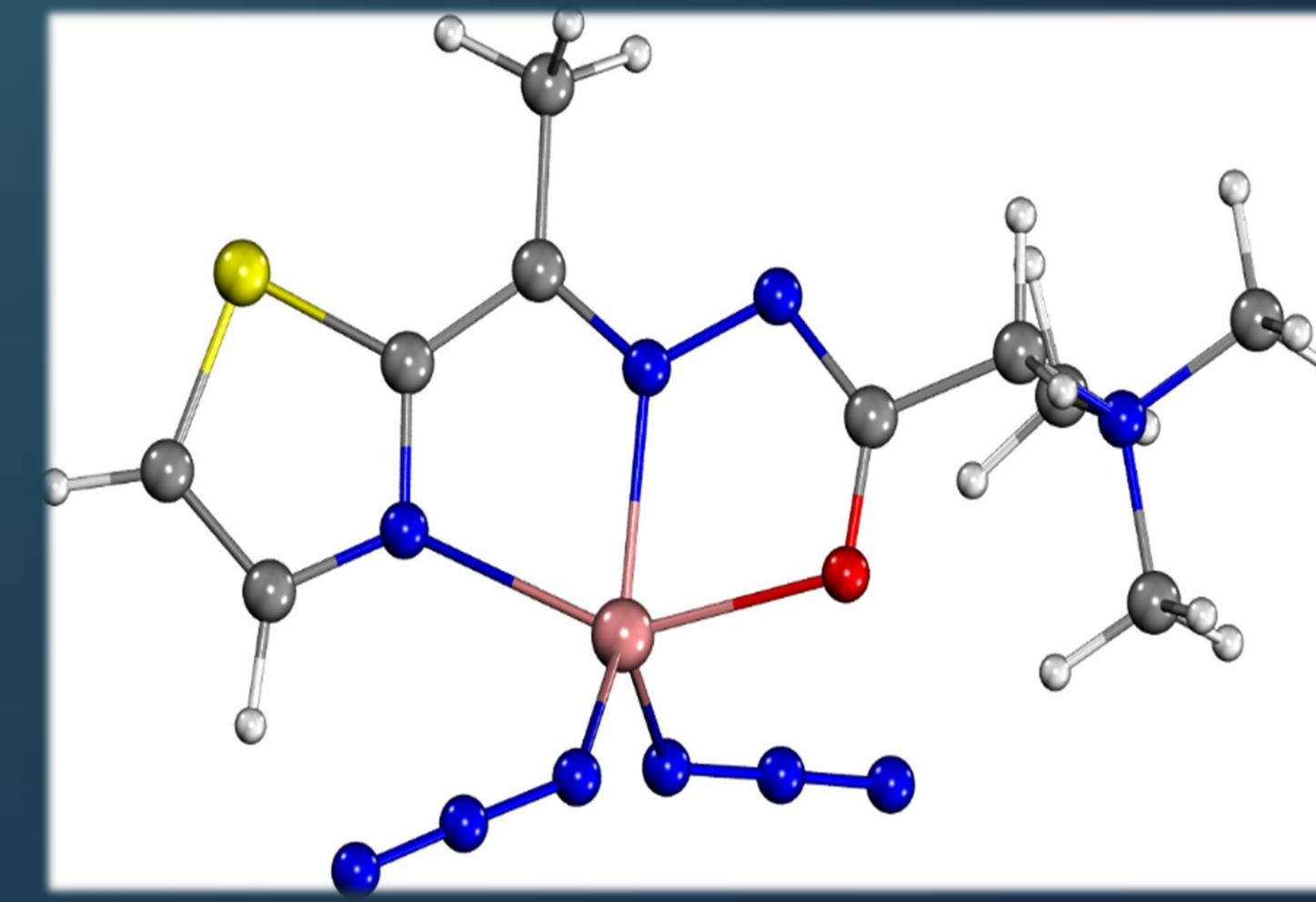


Fig. 5 Structure of  $[\text{MnL}^1(\text{N}_3)_2]$  complex optimized at ZORA-BP86-D3/TZP-COSMO(DMSO) level of theory