Serbian Young Chemists' Club

Serbian Chemical Society





Seventh Conference of the Young Chemists of Serbia Book of Abstracts

Belgrade, 2nd November 2019

CIP - Каталогизација у публикацији Народна библиотека Србије, Београд

54(048)(0.034.2) 577.1(048)(0.034.2) 60(048)(0.034.2) 66.017/.018(048)(0.034.2)

CONFERENCE of the Young Chemists of Serbia (7; 2019; Beograd)

Book of abstracts [Elektronski izvor] / Seventh Conference of the Young Chemists of Serbia, Belgrade, 2nd November 2019 ; [organized by] Serbian Chemical Society [and] Serbian Young Chemists Club ; [editors Tamara Todorović ... [et al.]]. - Belgrade : Serbian Chemical Society, 2019 (Belgrade : Development and Research Centre of Graphic Engineering Faculty of Technology and Metallurgy). - 1 elektronski optički disk (CD-ROM) ; 12 cm

Sistemski zahtevi: Nisu navedeni. - Nasl. sa naslovne strane dokumenta. - Tiraž 150. - Bibliografija uz većinu apstrakata. - Registar.

ISBN 978-86-7132-076-4

 а) Хемија -- Апстракти б) Биохемија -- Апстракти в) Биотехнологија -- Апстракти г) Наука о материјалима -- Апстракти

COBISS.SR-ID 280545292

7th CONFERENCE OF THE YOUNG CHEMISTS OF SERBIA BELGRADE, 2nd November 2019 **BOOK OF ABSTRACTS**

Published and Organized by

Serbian Chemical Society and Serbian Young Chemists Club

Karnegijeva 4/III, 11000 Belgrade, Serbia Tel./fax: +381 11 3370 467; www.shd.org.rs; office@shd.org.rs

Publisher

Vesna MIŠKOVIĆ-STANKOVIĆ, president of Serbian Chemical Society

Editors Tamara TODOROVIĆ Ljubodrag VUJISIĆ Jelena RADIVOJEVIĆ Vuk FILIPOVIĆ

Page Layout and Design **Vuk FILIPOVIĆ**

Circulation 150 copies

ISBN 978-86-7132-076-4

Printing

Development and Research Centre of Graphic Engineering Faculty of Technology and Metallurgy, Karnegijeva 4, Belgrade, Serbia

Year of Publication: 2019

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Supported by



Ministarstvo prosvete, nauke i tehnološkog razvoja Republike Srbije Ministry of Education, Science and Technological

Ministry of Education, Science and Technologica Development of Republic of Serbia



Evropska mreža mladih hemičara

The European Young Chemists' Network



Evonik Industries AG

IAP PP 01

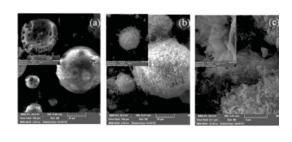
Biodiesel synthesis over green catalyst: The effect of thermal treatment of CaO/Zeolite precursor on catalytic activity

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The green CaO/Zeolite catalyst for methanolysis of fatty oils was synthesized entirely from the waste materials. CaO derived from chicken eggshell was loaded onto fly ashbased zeolite catalyst carrier by the wet impregnation method using an alcohol solution [1]. The effect of thermal activation at different temperatures ranging from 450 to 600 °C on catalytic activity was studied. The precursor and catalyst samples were characterized by XRD, FTIR, SEM, and Hg-porosimetry techniques. The catalytic tests were performed in a stirred batch reactor at the following reaction conditions: 60 °C - reaction temperature, 12:1 - methanol/oil molar ratio, and 4 wt% - catalyst concentration. The obtained results showed that the synthesized CaO/Zeolite catalyst has preserved alumosilicate framework-cancrinite type [1], with uniformly distributed calcium oxide (CaO) on its surface (Fig. 1c). It is shown that the catalyst sample calcinated at 550 °C exhibited the highest FAME content of 96.46%, which was achieved in 2 h (Fig. 2). Increasing temperature of calcination above 550 °C led to the formation of inactive calcium alumosilicate forms causing a decrease in the FAME content [2].



95.66 96.46 59.41 53.48 53.48 53.48 53.48 53.48 53.48 600 Calcination temperature (°C)

Figure 1. SEM microphages of (a) fly ash (b) Zeolite and (c) CaO/Zeolite-550 catalyst

Figure 2. The FAME content after 2 h with catalyst calcined at different temperatures.

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Acknowledgments

This study is supported by the Ministry od Education, Science and Technological Development of the Republic of Serbia (Project III45001).