

5TH REGIONAL
SYMPOSIUM ON
ELECTROCHEMISTRY
SOUTH EAST EUROPE

PROGRAM
BOOK OF ABSTRACTS

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PUBLISHED BY

Academician Evgeni Budevski Institute of Electrochemistry and Energy Systems
Bulgarian Academy of Sciences
10 Acad. G. Bonchev Str.
1113 Sofia, Bulgaria
Tel: (+359 2) 872 25 45
Fax: (+359 2) 872 25 44
www.bas.bg/cleps

PRINTED BY

Es Print
49 Kliment Ohridski Blvd.
1756 Sofia, Bulgaria

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DESIGN

Evgenia Ilkova

Sofia, Bulgaria
June 2015

ISBN 978-954-92483-4-0

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ELECTROCHEMISTRY
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June 7-11, 2015
RIU Pravets | Pravets, Bulgaria

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Sofia € 2015

Controlled Colloidal Synthesis and Basic Electrochemical Properties of TiO₂-Supported Pt

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TiO₂ was synthesized by forced hydrolysis process in order to be used as the support for Pt electrocatalyst (Pt/TiO₂). Pt was deposited from Pt colloid solution synthesized by microwave-assisted polyol process. TiO₂ powder (or C) was dispersed into H₂O and 2 M H₂SO₄. The obtained suspension was stirred and Pt colloid was added afterwards. Upon filtration and rinsing with water, the obtained Pt/TiO₂ (or Pt/C) catalyst was thermally treated at 160 °C in N₂ atmosphere. The catalyst water suspension for the preparation of 0.31 mg/cm² thin layer electrode was loaded onto glassy carbon disk. Physical and chemical properties of the TiO₂ and supported Pt catalyst were characterized by EDS, SEM, dynamic light scattering and XRD techniques. The influence of TiO₂ thermal treatment on the electrochemical properties of Pt/TiO₂ was also checked. Nominal Pt loading was 20 wt. %. EDS method revealed 18 wt. % loading of Pt on TiO₂. The electrochemical properties of the Pt/TiO₂ were examined by cyclic voltammetry in 0.1 M HClO₄. The obtained results were compared with those obtained for carbon-supported Pt under the same conditions (Figure 1). The charge corresponding to the hydrogen desorption on Pt/TiO₂ is lower than expected for 20 wt. % Pt. Although the voltammetric response is typical for Pt-based electrode material, the charge corresponds only to ~3 wt. % Pt. SEM images showed that TiO₂ particles tend to form 3µm-sized agglomerates of ellipsoidal shape. The results indicate the possibility for platinum particles to be trapped inside TiO₂ agglomerates. Even though calcination of TiO₂ improved the Pt uptake from colloidal dispersion, it is still not as effective as commercial carbon support.

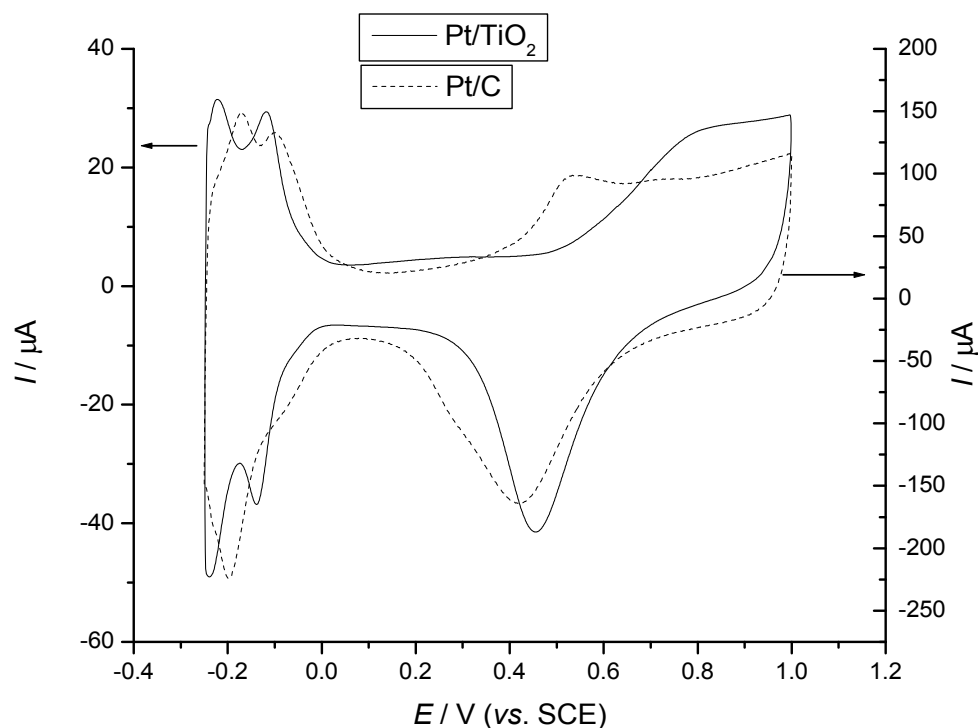


Figure 1. Cyclic voltammograms of Pt supported on thermally treated TiO₂ and Pt/C in deaerated 0.1 M HClO₄, sweep rate 50 mV s⁻¹.