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## BINARY MIXTURES OF SUBSTANCES WITH DOUBLE AND SINGLE BONDS

Milana Zarić<sup>1,2</sup>, Ivona Radović<sup>3</sup>, Mirjana Kijevčanin<sup>3</sup>

<sup>1</sup>Institute of Chemistry, Technology and Metallurgy, University of Belgrade – Belgrade, Serbia

<sup>2</sup>Centre of Excellence in Environmental Chemistry and Engineering – ICTM, University of Belgrade – Belgrade, Serbia

<sup>3</sup>Faculty of Technology and Metallurgy, University of Belgrade – Belgrade, Serbia

Corresponding author email: milana.zaric@ihm.bg.ac.rs

### Abstract

Studies of thermodynamic and transport properties (density, viscosity, refractive index and speed of sound) of pure substances and their mixtures give data on the behaviour of liquids that can be very important for processes in industry. In our previous work we studied pure substances cis-3-hexen-1-ol, 1-hexanol and n-hexane and their binary systems cis-3-hexen-1-ol + n-hexane and 1-hexanol + n-hexane. The results indicate double bond influence on the properties of the mixtures. In this work, we continue the study of the influence of double bonds on properties of mixtures in systems with double and single bonds. All thermodynamic properties measurements are reported in the range (288.15 to 318.15) K and at atmospheric pressure. Excess molar volume ( $V_e$ ), viscosity deviation ( $\Delta\eta$ ), refractive index deviation ( $\Delta nD$ ) and speed of sound deviation ( $\Delta u$ ) were calculated based on the experimental data. Derivate properties are fitted using Redlich-Kister equation. In order to additionally study molecular interaction, the FT-IR spectroscopy and quantum chemical calculations are performed. The comparison of systems with and without double bond shows the potential influence of double bond on thermodynamic properties of mixtures, that can be important in industry.

**Keywords:** thermodynamic properties, double bond, binary mixtures, FT-IR spectroscopy, intermolecular interactions.