

Serbian Ceramic Society Conference ADVANCED CERAMICS AND APPLICATION IX New Frontiers in Multifunctional Material Science and Processing

Serbian Ceramic Society Institute of Technical Sciences of SASA Institute for Testing of Materials Institute of Chemistry Technology and Metallurgy Institute for Technology of Nuclear and Other Raw Mineral Materials

PROGRAM AND THE BOOK OF ABSTRACTS

Serbian Academy of Sciences and Arts, Knez Mihailova 35 Serbia, Belgrade, 20-21. September 2021. Serbian Ceramic Society Conference ADVANCED CERAMICS AND APPLICATION IX New Frontiers in Multifunctional Material Science and Processing

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INV Influence of the water flow lens system on performances of the different laboratory made Sb₂S₃-based and commercial solar cells

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Here, the behavior of different types of solar cells at a low light intensity, measured with and without using the water flow lens (WFL) system is investigated. This system enables the cooling of the surface of the solar cell/modules/panels, indirectly cooling the surrounding, and allows investigating of the influence of higher or lower intensities of the light with the inevitable change in the spectrum. All of these effects are very important and can greatly contribute to the better photovoltaic performance of the observed cells. In this study, laboratory-made and commercial solar cells were studied at 5 % sun and (or) 35 % sun using a tungsten and halogen lamp, respectively. Comparing the obtained results performed when the WFL system is used and left out, it was confirmed that the WFL system facilitates obtaining better photovoltaic properties for all investigated solar cells.

INV

Cobalt-based catalysts in catalyticoxidationoftartrazine activated by Oxone®

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Two different types of cobalt-based catalysts were synthesized and tested as Oxone[®] activators in catalytic oxidation of model dye, tartrazine. First type of catalyst was cobalt impregnated aluminum pillared montmorillonite, and the second one was cobalt-doped alumina. Aluminum pillared montmorillonitewas synthesized from Na-exchanged Wyoming clay and impregnated with cobalt using incipient wetnessimpregnation method. Cobalt-doped alumina catalysts were synthetized using the sol-gel method. Three calcination temperatures were employed: 500 °C, 1000 °C and 1100 °C. The degradation of tartrazine was monitored using Thermo Scientific Evolution 220 UV–Visible Spectrophotometer in the wavelength range from 200–600 nm.In this wavelength range the monitoring of decolorization, along with registering the emergence, followed by degradation,ofdetectable degradation products was achieved.The effect of the different reaction parameters on decolorization and degradation efficiency was tested, includingtheinfluence of the mass of catalyst, reaction temperature and initial pH. It was found that the increase of temperature and the mass of

catalysts were beneficial for the reaction.All investigated catalysts were found to be very efficient in the Oxone® initiated tartrazine decolorization and degradation.

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INV

Preparation and Characterization Of Active Carbon Microspheres Obtained From Fructose And Adsorption Application

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Carbon material with active surface properties have been synthesized by hydrothermal method from fructose using 40% and 80% phosphoric acid (H₃PO₄) solution, at temperature of 260°C and fructose concentration of 2M. The aim of this investigation was active carbon material synthesis which is completed by one step reaction, which was not the case in our previous works. Thus, compared with other samples from our works, this way is more economic and faster since both reaction of carbonization and activation was finished in one step. The hydrothermal process, in general, includes heat treatment of carbohydrate solutions under autogenous pressure at low temperatures (150-260 °C). Obtained solid carbon material has uniform morphology, amorphous structure and high content of oxygen functional groups. Prepared active carbon material is made up of spherical microsphere particles with the diameter in the range of 0.6-2.7 µm. The morphology and surface properties of obtained material were characterized by scanning electron microscopy (SEM), Fourier-transform infrared (FTIR) spectra. Adsorption and desorption isotherms of N₂ were measured on carbon containing material at -196 °C using the gravimetric McBain method. Adsorption from aqueous solutions of Methylene Blue (MB) onto prepared carbon material was conducted by changing concentration of MB from 200-500mg/dm³. The best fit of the kinetic results was achieved by a pseudo second-order equation. Also, this nature of material is applicable in other systems regarding environmental protection and dye pollution prevention.