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## **Poboljšanje oksidacije etanola na Pd elektrohemijski nataloženom preko $\text{Sb}_2\text{O}_3$**

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Elektrohemijski nataloženi Pd i Sb-Pd katalizatori su ispitivani u reakciji oksidacije etanola cikličnom voltametrijom i hronoamperometrijskom metodom. Paladijum-antimon je sintetisan na nosaču od staklastog ugljenika u dvo-stepenom procesu, tako što je najpre nataložen Sb, a zatim Pd. Tako je dobijen Sb-Pd elektrokatalizator sa atomskim odnosom Sn:Pd (0.2:0.8). U odnosu na nemodifikovanu Pd elektrodu, dodatak Sb rezultuje u povećanoj aktivnosti i stabilnosti bimetalnog katalizatora u elektrohemijskoj oksidaciji etanola. Posle hronoamperometrijskih ispitivanja, Sb-Pd katalizator je podvrgnut cikliziranju kako bi se povratila aktivnost, čime je utvrđena stabilnost sastava elektrode. Poboljšane karakteristike Sb-Pd su pripisane bifunkcionalnom mehanizmu delovanja i elektronskom efektu.

## **Enhancement of ethanol oxidation on Pd electrodeposited over $\text{Sb}_2\text{O}_3$**

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Electrodeposited Pd and Sb-Pd catalysts were tested for the ethanol oxidation reaction by cyclic voltammetry and chronoamperometry. Palladium-antimony deposit was synthesized on a glassy carbon substrate by a two-step process, comprising deposition of Sb followed by deposition of Pd. The Sb-Pd electrocatalyst with the Sn:Pd atomic ratio 0.2:0.8 was obtained. Compared with pure Pd, the incorporation of Sb results in the increase of activity and stability of bimetallic catalyst for the electrocatalytic ethanol oxidation reaction. Besides upon the end of the current-time transient, the investigated Sb-Pd catalyst was subjected to the potential cycling showing the ability to recover activity loss implying the surface composition stability. The enhanced performance of Sb-Pd is mainly ascribed to the bifunctional mechanism and electronic effect.

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