

## Book of Abstracts

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## Synthesis and characterization of octahedral Ni(II) complex with condensation product of 2-acetylthiazole and thiosemicarbazide

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The ligand (**HL**) was synthetized by the reaction of 2-acetylthiazole and thiosemicarbazide in molar ratio 1:1 in water. In the reaction of ligand (**HL**) and nickel(II) tetrafluoroborate hexahydrate  $[Ni(\mathbf{HL})_2](BF_4)_2$  complex was obtained. The Ni(II) ion has octahedral coordination geometry. The tridentate ligand (**HL**) is coordinated to the nickel ion with a NNS set of donor atoms forming two five-membered chelate rings. The ligand (**HL**) and the  $[Ni(\mathbf{HL})_2](BF_4)_2$  complex were characterized by elemental analysis, IR and UV/Vis spectroscopy and the structure of the complex was defined by X-ray analysis.

Figure 1. Synthesis of  $[Ni(HL)_2](BF_4)_2$  complex.

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