= DISCUSSIONS =

Reply to the Comment by Silvano Romano [1] on the Paper [2]¹

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It is true that this totally open ensemble (which is not-generalized ensemble) had already been discussed in the literature, but never with this specific derivation. This derivation is direct, it starts from the discussion of N, V, E reservoirs, thus being especially suitable for physicochemists. Our paper also contains a suitable table of partition sums and thermodynamic potentials.

Our results coincide with that given in the literature [3]. Starting from the grand canonical ensemble, we can derive the statistical sum of a totally open system using the second postulate of statistical thermodynamics [3]:

$$\Pi(T, P, \mu) = \sum_{i} \Xi(T, V, \mu) \exp\left(\frac{-PV_{i}}{k_{\rm B}T}\right),$$

where Π is the totally open partition sum and Ξ the grand canonical partition sum:

$$\Xi(T,V,\mu) = \sum_{i} \sum_{k} \exp\left(\frac{\mu N_{k} - E_{i}}{k_{\mathrm{B}}T}\right).$$

Thus, the statistical sum of a totally open system is:

$$\Pi(T, P, \mu) = \sum_{i} \sum_{j} \sum_{k} \exp\left(\frac{\mu N_{k} - E_{i} - PV_{j}}{k_{\mathrm{B}}T}\right),$$

and if considering the degeneration or statistical weight, the previous term becomes:

$$\Pi(T, P, \mu) = \sum_{i} \sum_{j} \sum_{k} g(E_{i}, V_{j}, N_{k})$$
$$\times \exp\left(\frac{\mu N_{k} - E_{i} - PV_{j}}{k_{B}T}\right).$$

REFERENCES

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- I. M. Stanković, V. M. Marković, and Lj. Z. Kolar-Anić Russ. J. Phys. Chem. A. 85 (13), 2257 (2011).
- 3. A. Münster, *Statistical Thermodynamics* (Springer, Berlin Heidelberg New York, Academic Press, New York, London, 1969), vol. 1, ch. 3, sects. 3.1 and 3.2.

¹ The article is published in the original.